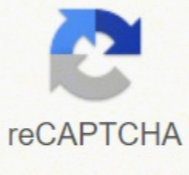
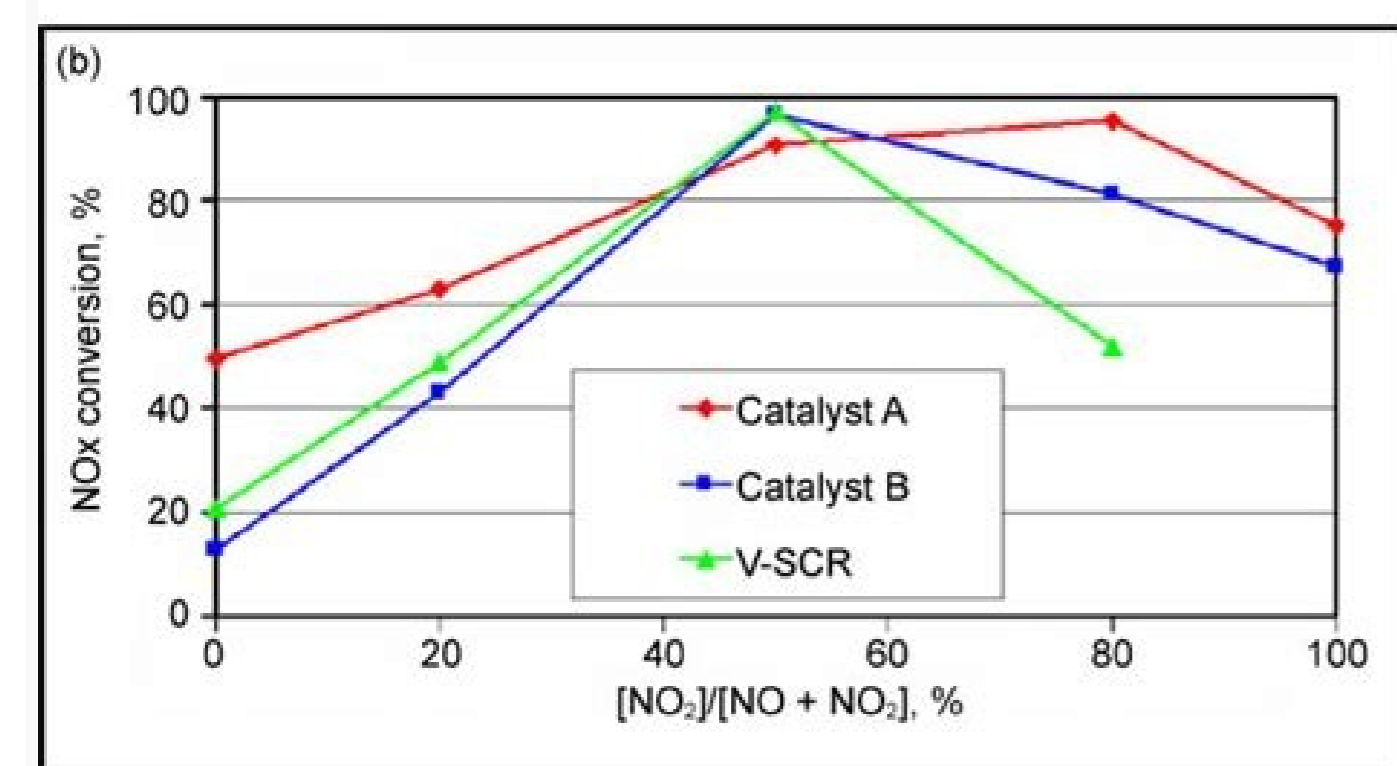
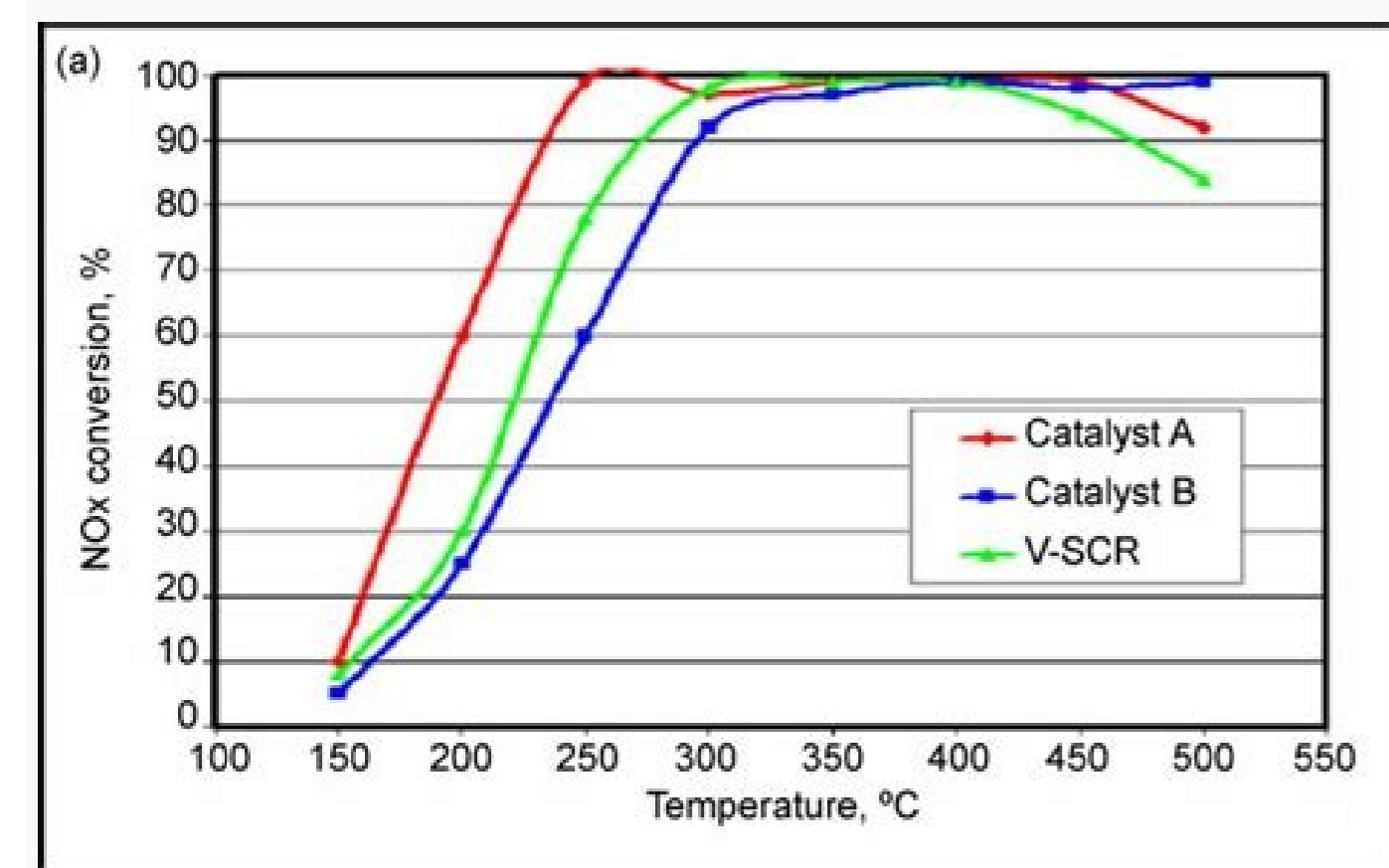
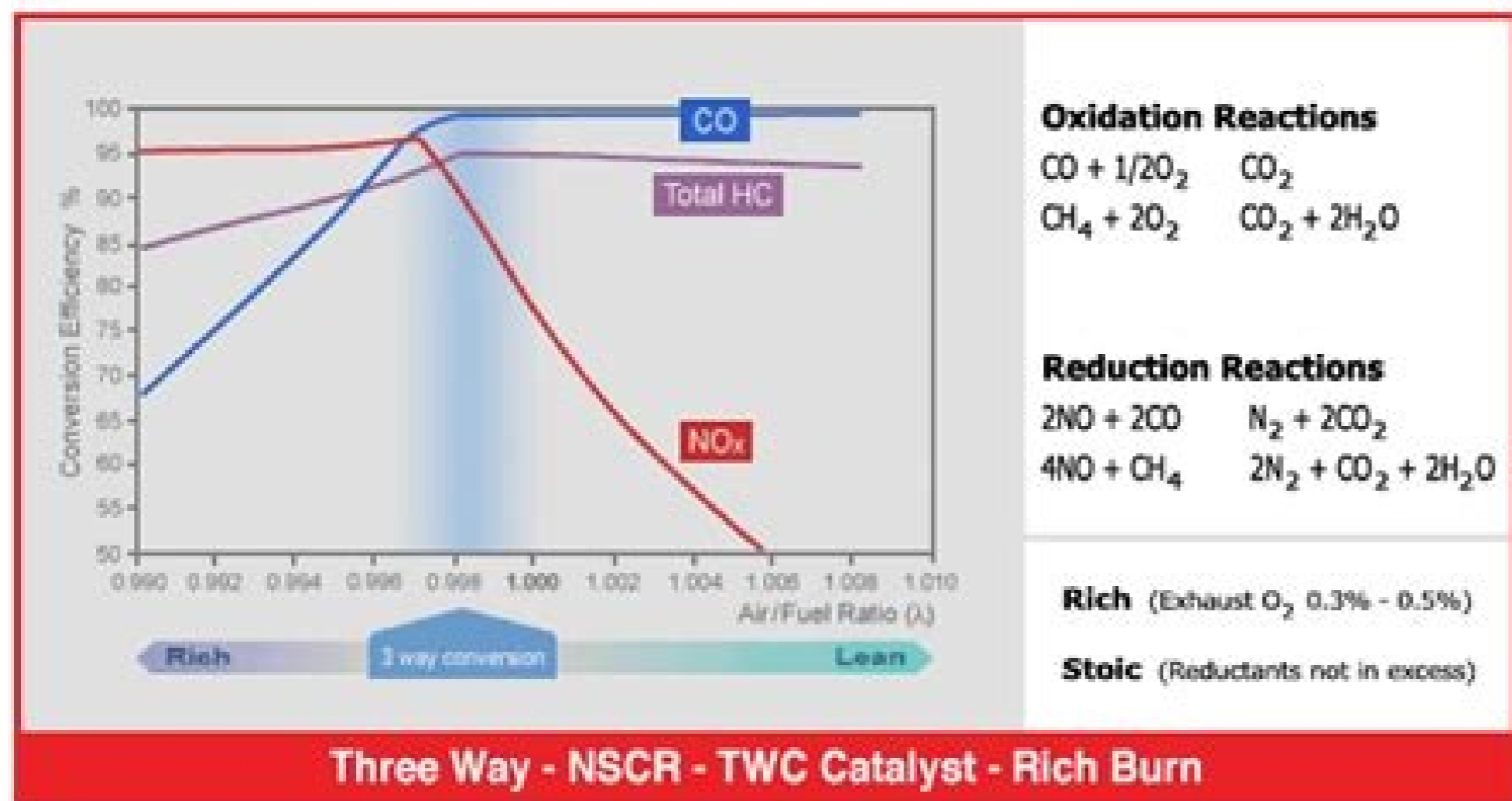
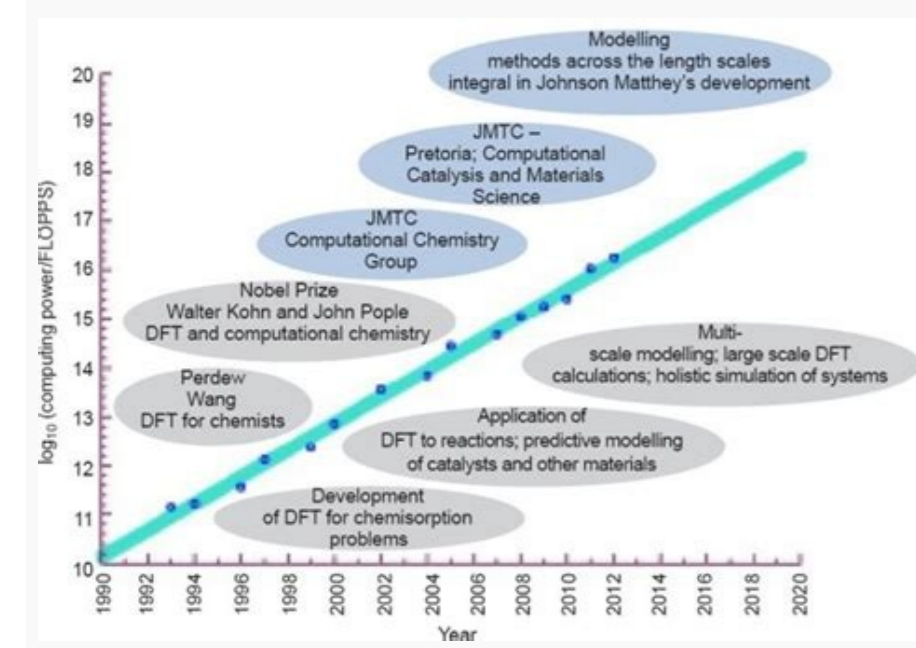




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Alcohol Dehydrogenase 132

Catalogue Number:

ADH-132

General Properties

Description:

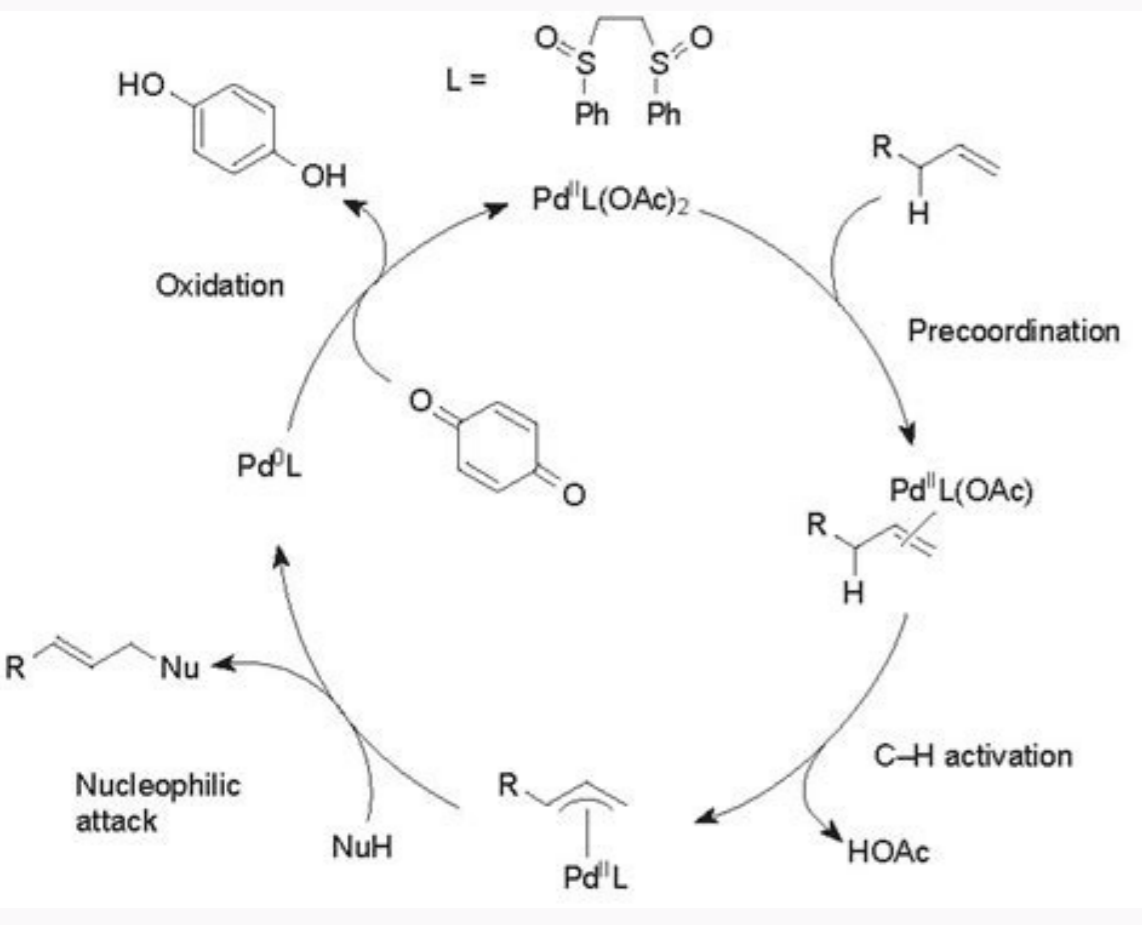
NADP dependent glycerol dehydrogenase, member of the aldo-keto reductase superfamily. Its substrate spectrum includes a variety aldehydes and α -diketones.

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Platinum Metals Rev., Å Rayna160; 2001,ÅÅ MENINA 160; 45,Å MENIA GIRL 160;(3),Å MENIA,110The Guide to Reaction Catalâ@m contains information about the 106 important reactions in industrial chemical. After selecting the reaction of interest, the pointer inside the bar insert lines up with details of the recommended catalyst and reaction conditions. Comments on the reaction and suggestions for improvements are listed on the pages turned. Details of hydrogenation, coupling, selective oxidation, dehydrogenation, hydroformy, carbonization, decalazation, hydrosililaâo and cyclopropanization reactions are presented and suggestions for Johnson Matthey catalysts are made. Chemical Johnson Mattheyvado Ahem Ahem.com., Catalytic Reaction Guide, laminated, multi-pocket folded Slide-Chart with flaps, has a wealth of information about catalytic reactions, the chemical products used to produce them, the 106 equas involved, with comments and specifications for each. Call us today at 1.800.323.4433. Å Joshua 160; Or fill in our free custom design form and we'll create a stowerve tailored to your needs! Visit us in the america ©.com/slide-charts for more creative marketing tools! Watch The Johnson Mathey Slide Chart in a move! Platinum Metals Rev., Å To help industrial users identify the most appropriate catalyst for more than sixty different processes, Johnson Matthey has recently prepared a guide that will contribute to the most effective use of platinum group metal catalysts. When a pointer in the slide screen is © moved to indicate the desired product or reagent, the recommended heterogâ©neo or homogâ©neo catalyst and the process conditions are presented; Brief observations© are also presented. The Catalmatics Reaction Guide is now available in English, French and France, and can be obtained in writing for Johnson Matthey's office on Orchard Road, England; Paris, France; or West Germany. 10-Feb-2015 Provides mobile access to catalyst recommendations over more than 150 Johnson Matthey Catalysis and Chiral Technologies (JMCCT) business reactions, a business unit of the Johnson Matthey Fine Chemical Division, launched your Catalysa Reaction Guide (CRG) as a free mobile application. The application provides cellular access to catalyst recommendations over more than 150 chemical reactions in the area of homogâ©nea, heterogâ©nea and © acrogâ©ta. Users can easily view and search a complete list of products with simplife©d technical support, take advantage of an efficient quote request process, and get contact information for the company's full service offering. JMCCT says it has made it the mobile guide to allow customers to make informed purchasing decisions anywhere in the world. In addition© to the catalyst recommendations, the application presents a list of chemical abstracts service (CAS) searchable products and relevant bibliographic references. The interface is © fully customized through a user profile that synchronizes data with other devices for easy follow©up of recommendations and details. The CRG Application is © a cutting-edge and efficient means for customers to access catalytic reaction data Gerard Compagnoni, Managing Director of Johnson Matthey Catalysis and Chiral Technologies, said: "The CRG application is © cutting-edge and efficient means for customers to access catalytic reaction data, and take advantage of the technical © of jmcctâs on the go, we are very excited to present the application in benefit of our customers, and look forward to debuting soon the online version of CRG to complete this feature" The migrate to a digital platform was directly driven by crg's demand from customers in a more controllable format than traditional printing, and was formally completed and put into testing At the end of last year. Customers can download the application for free on any iPhone, iPad or Android device. A supplementary online CRG will enter into site this spring. Companies The Catalytic Reaction Guide (CRG) is © a free useful tool designed to provide customers with access to catalyst recommendations over more than 180 reactions in homogeneous, heterogânea and so©picking and so©atric water reares. It allows users to easily browse our complete product lists. The mobile app includes features like âf" to help answer your technical questions© and ask for a quote âf" to help speed up your purchasing process. A printed version © is available to keep working. Platinum Metals Rev., 2001, 45, (3), 110Johnson Matthey have published an informative brochure of 82 pages, â©The CatalystrÅ Technical Manual, which covers the use of catalysts for important chemical reactions in industrial sane. The manual recommends homogenous, heterogenous and FibreCat™ catalysts anchored homogeneously for each type of reaction discussed. This allows the user to make informed decisions relevant to their individual circumstances. © is also included the advice of optimizing the operating conditions. The sections cover hydrogenation, dehydrogenation, hydroformy, carbonization, decarbonization, hydrosylled, cross-coupling, cycling and carbene reaction, isomerizaâ-o, oligomerization and polymerization, selective oxidation, chiral catalysts and also © particulate catalysts. The brochure has extensive information on therapeutic reaction schemes and advice on the handling and service of catalysts such as recycling. To obtain âThe© Technical Catalyst Manualâ, â The Catalmatic Samatic Reaction Guide âf or a Catalyst Sample Kit, please contact Chemical Sales, First Floor â HO Building, Johnson Matthey, Orchard Road, Royston, Herts SG8 5HE; Fax: +44 (0) 1763 253573; E-mail: chemicals@matthey.com; Internet: . . . Indeed for many years the metals company Johnson Matthey operated a loan scheme where they would give aspiring researchers 100 g or so of ruthenium trichloride to experiment with in the hope the chemists would find new uses for the material. The loans scheme operated for the pricier metals like rhodium as well, but only in little 5 g pots. Oct 08, 2019 · The demand for fossil derivate fuels and chemicals has increased, augmenting concerns on climate change, global economic stability, and sustainability on fossil resources. Therefore, the production of fuels and chemicals from alternative and renewable resources has attracted considerable and growing attention. Ethanol is a promising biofuel that can reduce ... Platinum is a chemical element with the symbol Pt and atomic number 78. It is a dense, malleable, ductile, highly unreactive, precious, silverish-white transition metal.Its name originates from Spanish platina, a diminutive of plata "silver.". Platinum is a member of the platinumum group of elements and group 10 of the periodic table of elements.It has six naturally occurring isotopes. Johnson Matthey perfected the techniques of separating and refining the platinum group metals and in 1879 Matthey produced a standard metre measure made of a platinum and iridium alloy. Platinum compounds have been well documented, perhaps none more so than cis-diamminedichloroplatinum(II), cisplatin. Histoire. Le palladium a été découvert, avec le rhodium, à partir du platine natif brut en 1803 par le médecin et chimiste anglais William Hyde Wollaston [7], [8].Le nom, donné par Wollaston lui-même l'année suivant sa découverte, dérive de l'astéroïde Pallas découvert le 28 mars 1802 par l'astronome allemand Heinrich W.M. Olbers [9].Le nom de l'astéroïde fait référence à la ... Reaction engineering deals with the holistic design of reactions and appropriate reactors, using the concepts of materials and energy balances, kinetics and chemical thermodynamics. These concepts apply across all sectors of chemical engineering, from petrochemicals to food, from pollution control to biotechnology. Palladium is a chemical element with the symbol Pd and atomic number 46. It is a rare and lustrous silvery-white metal discovered in 1803 by the English chemist William Hyde Wollaston. He named it after the asteroid Pallas, which was itself named after the epithet of the Greek goddess Athena, acquired by her when she slew Pallas.Palladium, platinum, rhodium, ruthenium, iridium ... The Oxygen Evolution Reaction (OER) at the anode is the rate determining step of the catalysed electrolysis reaction. For this we offer high activity iridium products, including Iridium black - a durable and high surface area catalyst.. Iridium plays a crucial role in the PEM electrolysis and the future hydrogen economy.

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